

**effects of intermolecular forces intermolecular and** - optional experiment the effects of intermolecular forces part text 2 aim to investigate surface tension and to determine the relation between surface tension and intermolecular forces, **effects of increase on intermolecular forces strength** - summary experiment 9 is conducted to identify the importance of intermolecular forces and how they affect molecules intermolecular forces are forces between molecules that determine whether the molecule is a solid liquid or gas under standard conditions, **school effect of intermolecular forces on solubility** - the intermolecular forces of the solids and solvents the 1s are the solute solute forces the 2s are the solvent solvent forces and the 3s are the solvent solute forces in the capillary experiment of this lab it was concluded that water reacted more with glass than tygon and hexane reacted more with tygon than glass, **the effect of intermolecular forces on surface tension** - and conducting an experiment examining the effect of intermolecular forces on surface tension write a laboratory report that explains your procedures and results and confirms or rejects your hypothesis, **lab 17 the effect of intermolecular forces on** - methanol ch 3 oh and ethanol c 2 h 5 oh are two of the alcohols that we will use in this experiment you will examine the molecular structure of alkanes and alcohols for the presence and relative strength of two intermolecular forces hydrogen bonding and dispersion forces due to mass of a molecule, **intermolecular and interatomic forces intermolecular** - the effects of intermolecular forces the following five experiments investigate the effect of various physical properties evaporation surface tension solubility boiling point and capillarity of substances and determine how these properties relate to intermolecular forces each experiment will look at a different property, **effects of increase on intermolecular forces strength** - intermolecular forces are forces between molecules in a compound alcohol s intermolecular forces are van deer wall forces and hydrogen bonds this experiment is designed to determine whether the increase in relative molecular mass  $m_r$  in alcohols methanol ethanol and propanol will increase the strength of the intermolecular forces specifically the van deer wall forces, **intermolecular and interatomic forces intermolecular** - the effects of intermolecular forces the following five experiments investigate the effect of various physical properties evaporation surface tension solubility boiling point and capillarity of substances and determine how these properties relate to intermolecular forces each experiment will look at a different property

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